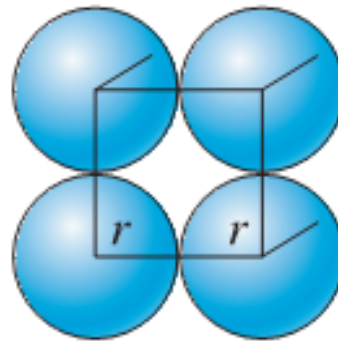


Odrediti vezu između atomskog radijusa i parametara rešetke za SC, BCC, FCC

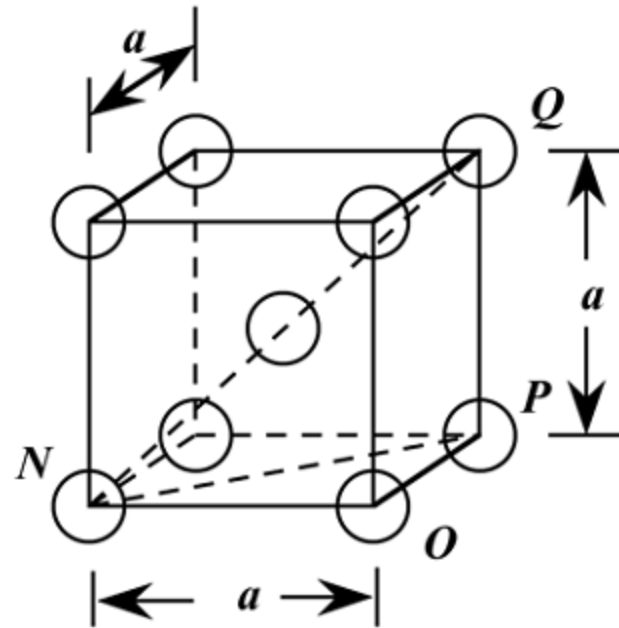
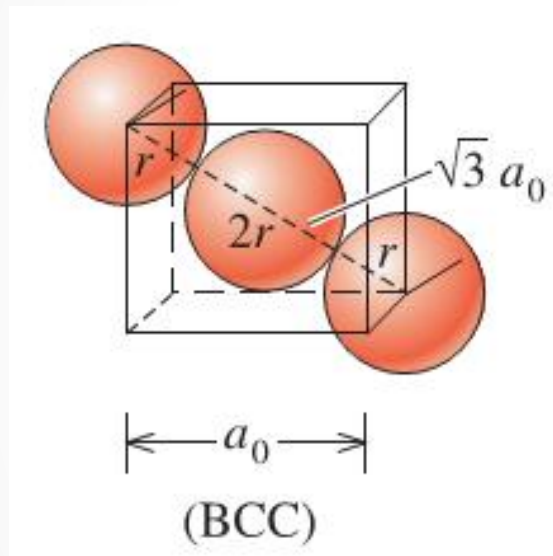


$$| \leftarrow a_0 \rightarrow |$$

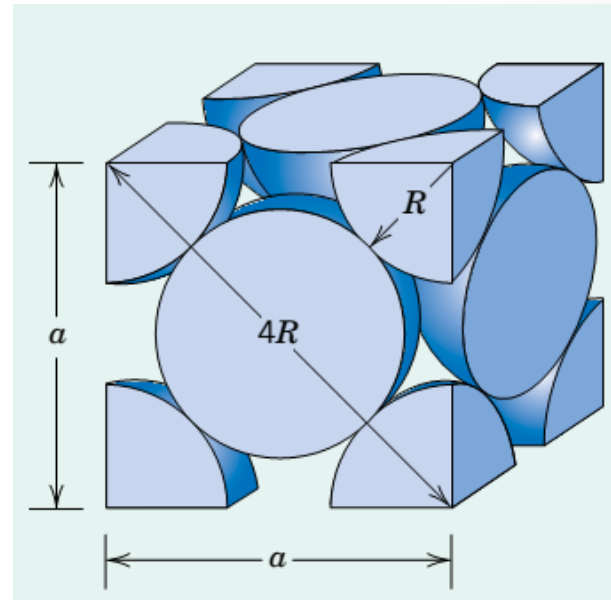
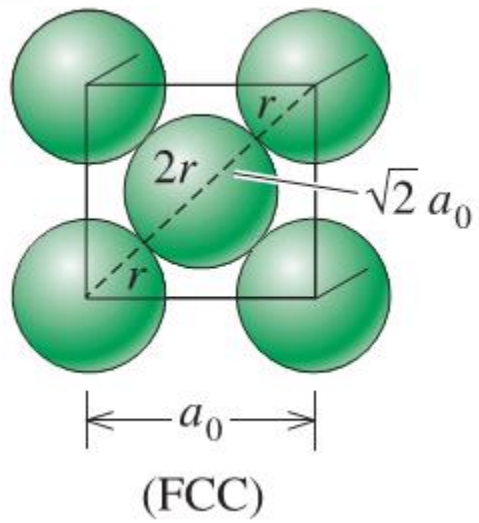
(SC)

$$a_0 = 2r$$

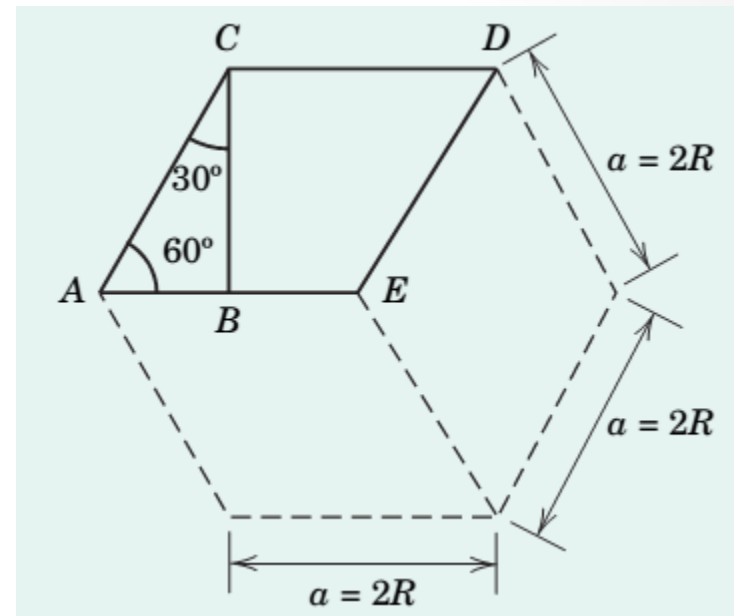
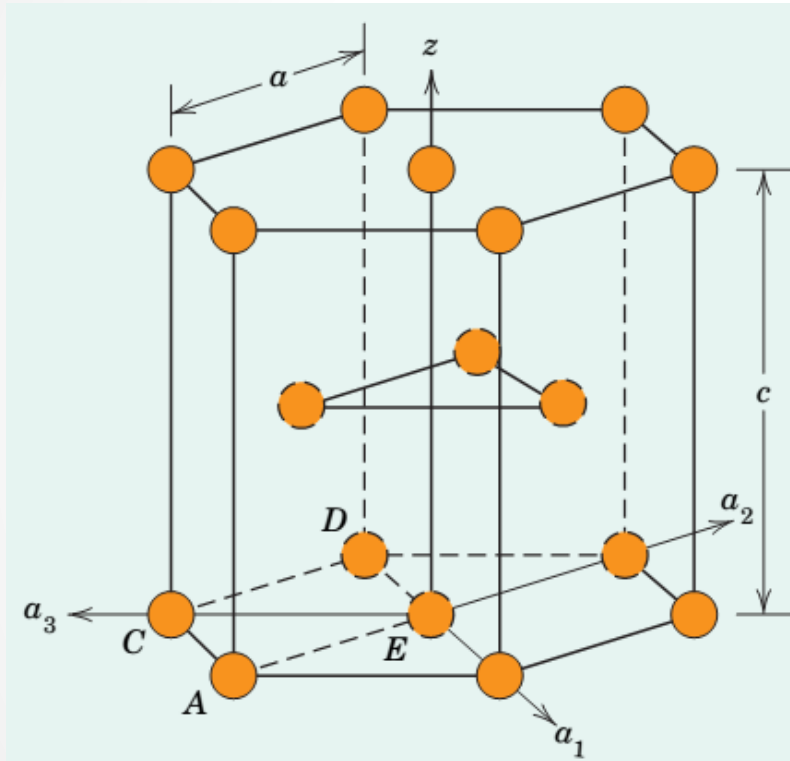
BCC



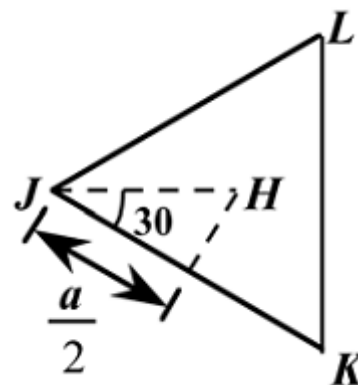
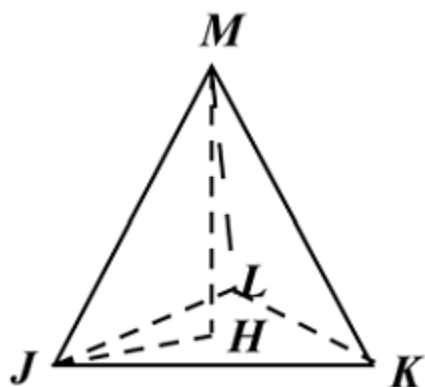
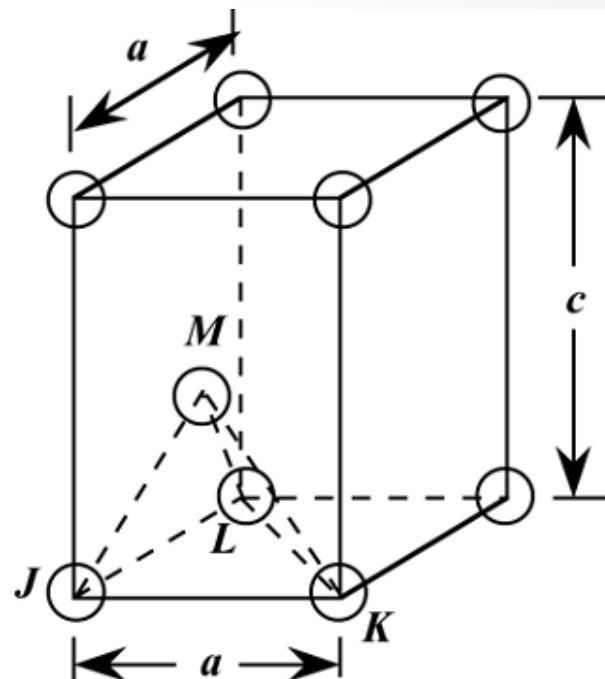
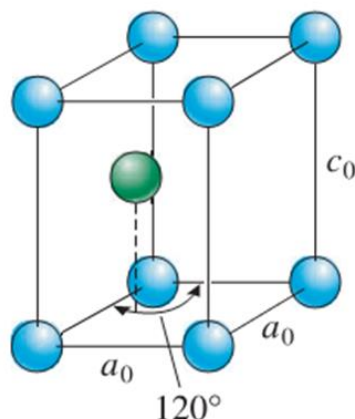
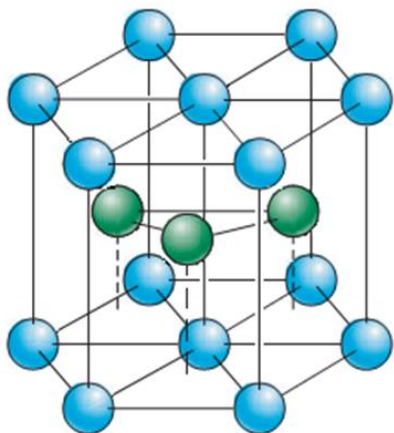
FCC



HCP



Idealan c/a odnos kod HCP rešetke



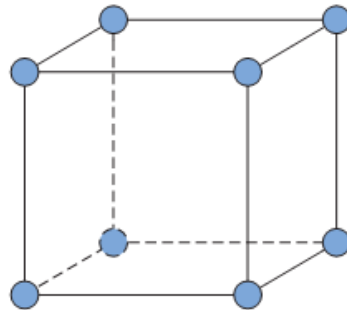
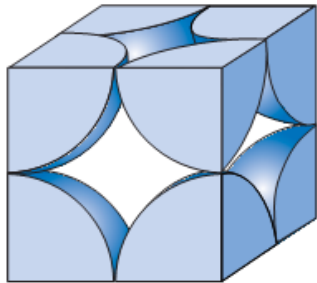
Faktor pakovanja

- Faktor pakovanja predstavlja udeo prostora koji zauzimaju atomi u jed. ćeliji pod uslovom da su atomi čvrste sfere takve veličine da dodiruju svog najbližeg suseda.

$$\text{Faktor pakovanja} = \frac{(\text{br. atoma/ćeliji})(\text{zapremina svakog atoma})}{\text{zapremina jed. ćelije}}$$

Broj atoma po jed. ćeliji

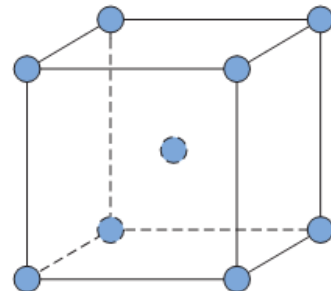
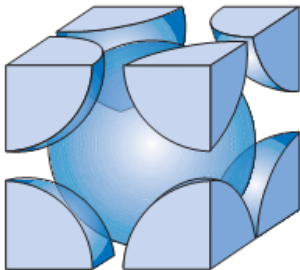
SC



Atoma po ćeliji

1

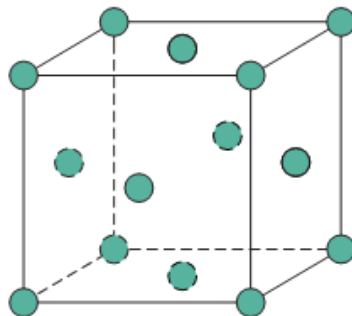
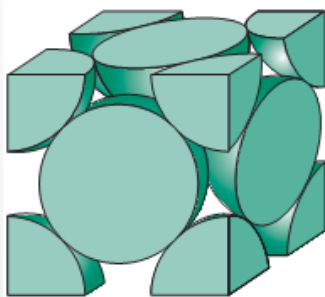
BCC



Atoma po ćeliji

2

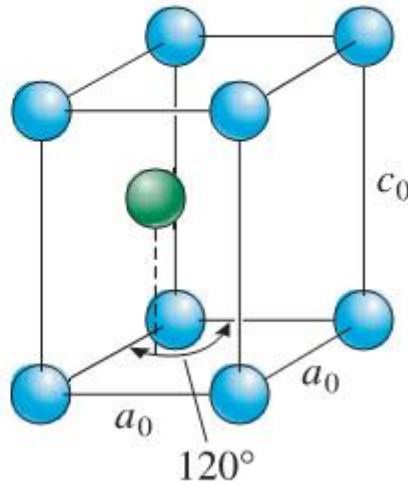
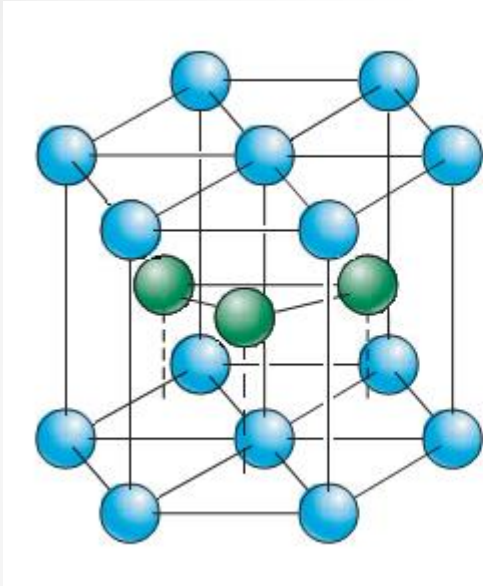
FCC



Atoma po ćeliji

4

Broj atoma po jed. ćeliji



Atoma po ćeliji

$$3 \times 2 = 6$$

HCP je kao 3 BCC!

Faktor pakovanja

- Faktor pakovanja predstavlja udeo prostora koji zauzimaju atomi u jed. ćeliji pod uslovom da su atomi čvrste sfere takve veličine da dodiruju svog najbližeg suseda.

$$\text{Faktor pakovanja} = \frac{(\text{br. atoma/ćeliji})(\text{zapremina svakog atoma})}{\text{zapremina jed. ćelije}}$$

Struktura	Fak. pak.
Prosta kubna (SC)	0.52
Prostorno. centr. kubna (BCC)	0.68
Povr. cent. kubna (FCC)	0.74
Heksagonalna blisko pakovana (HCP)	0.74